

Tetrahedron Letters, Vol. 35. No. 21, pp. 3571-3574, 1994 Elsevier Science Ltd Printed in Great Britain 0040-4039/94 \$7.00+0.00

0040-4039(94)E0590-T

A New Reductive Acidolysis Final Deprotection Strategy in Solid Phase Peptide Synthesis. Use of a New Safety-Catch Linker¹

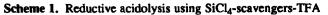
Yoshiaki Kiso,* Toshio Fukui, Shigeki Tanaka, Tooru Kimura and Kenichi Akaji

Department of Medicinal Chemistry, Kyoto Pharmaceutical University, Yamashina-ku, Kyoto 607, Japan

Abstract: A new reductive acidolysis final deprotection strategy in solid phase peptide synthesis was developed using a new safety-catch linker; this new strategy was based on a two-dimensional protection scheme employing acid-labile temporary and acid-stable but reductive acidolysis-cleavable semipermanent protecting groups.

In a two-dimensional protection scheme of peptide synthesis, semi-permanent protecting groups must be entirely stable under synthetic conditions including selective deprotection of temporary protecting groups and completely cleavable at the final deprotection stage. Recently, a series of safety-catch protecting groups have been developed on the basis of 4-methylsulfinylbenzyl (Msob) group.²⁻⁹ These protecting groups are stable to acid because of electron-withdrawing character of the sulfoxide. In a one-pot reaction using tetrachlorosilanescavengers-trifluoroacetic acid (TFA) system, the Msob-derived protecting groups are smoothly reduced to the corresponding sulfide-form and then cleaved by acidolysis (reductive acidolysis, Scheme 1).^{2,5-9} These groups are suitable as semi-permanent protecting groups in solid phase peptide synthesis (SPPS)¹⁰, since the

$$R-O-CH_2 \longrightarrow \stackrel{O}{\mathbb{S}}^{-}CH_3 \longrightarrow \left[R-O-CH_2 \longrightarrow S-CH_3 \right] \longrightarrow R-OH$$



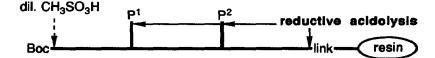
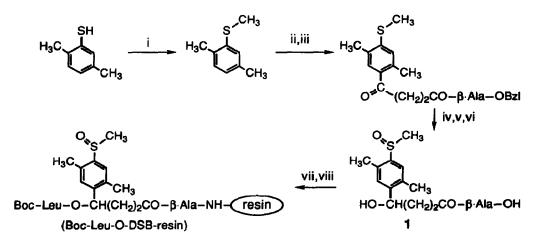


Fig. 1. A two dimensional protection scheme for SPPS by a reductive acidolysis final deprotection strategy

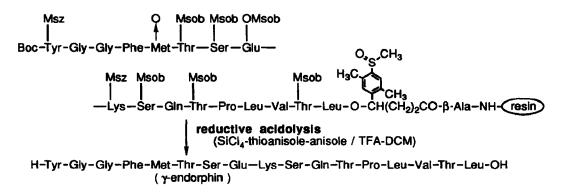
Msob-type groups possess high stability during repetitive acid deblocking cycles, but are cleavable with mild reductive acidolysis.

In this communication, we describe a new two-dimensional protection scheme in SPPS employing the safety-catch linkage to the resin and reductive acidolysis final deprotection strategy. This two-dimensional protection scheme is based on a combination of an acid-labile temporary group [Boc (*t*-butoxycarbonyl) in Fig. 1] and acid-stable but reductive acidolysis-labile semipermanent protecting groups (P^1 , P^2 and anchoring link in Fig. 1).

In order to introduce a safety-catch type ester linkage to aminomethylated-polystyrene resin, we designed a new handle reagent, 4-(2,5-dimethyl-4-methylsulfinylphenyl)-4-hydroxybutanoic (DSB) acid. Using this reagent, the DSB-resin was prepared according to the scheme shown in Scheme 2. Friedel-Crafts reaction of succinic anhydride and 2,5-dimethylthioanisole prepared from the corresponding thiophenol gave a γ -



Scheme 2. Preparation of DSB-resin. *Reagents*, i, CH₃I; ii, succinic anhydride, AlCl₃; iii, H-β-Ala-OBzl, DCC; iv, NaOH; v, NaBH₄; vi, H₂O₂; vii, NH₂-resin, disopropylethylamine, BOP; viii, Boc-Leu, DIPCDI, DMAP.



Scheme 3. Synthesis of γ -endorphin by a reductive acidolysis final deprotection strategy

ketocarboxylic acid. After the coupling with β-Ala-OBzl and saponification, the ketone moiety was reduced to alcohol using NaBH₄. The oxidation of the resulting sulfide to sulfoxide using H₂O₂-AcOH gave a hydroxycarboxylic acid 1 in 47.5 % yield from the corresponding thiophenol. The handle reagent 1 was introduced to an aminomethylated polystyrene-resin by benzotriazol-1-yloxytris(dimethylamino)phosphonium hexafluorophosphate (BOP)-activation, and the amount of incorporated β-Ala was determined by amino acid analysis of the resulting resin. The C-terminal amino acid was coupled with DSB-resin by diisopropylcarbodiimide (DIPCDI)-*N*,*N*-dimethylaminopyridine (DMAP) method. Using Boc-Leu-O-DSB-resin thus obtained, the stability of the anchoring linkage was examined. The amount of cleavage by TFA-anisole (25 °C, 24 h) was 3.5 %, whereas 90.5 % of loaded amino acid was cleaved by reductive acidolysis [SiCl₄-thioanisole-anisole-TFA (25 °C, 3 h)]. The amount of D-Leu in the liberated amino acid was 1.7 % when determined by GITC method.¹¹

In order to demonstrate the usefulness of the new strategy and DSB-resin, we synthesized γ -endorphin (H-Tyr-Gly-Gly-Phe-Met-Thr-Ser-Glu-Lys-Ser-Gln-Thr-Pro-Leu-Val-Thr-Leu-OH,¹² Scheme 3). Starting from Boc-Leu-O-DSB-resin, the protected peptide-resin was prepared according to the schedule of the efficient SPPS¹³ using N^{α}-Boc-amino acids bearing reductive acidolysis-cleavable side chain protecting groups, *i.e.* Tyr(Msz),⁶ Met(O), Thr(Msob),⁶ Ser(Msob),^{4,6} Glu(OMsob)⁶ and Lys(Msz)⁵ (Msz=4-methylsulfinyl-benzyloxycarbonyl). The protected γ -endorphin-resin [Boc-Tyr(Msz)-Gly-Gly-Phe-Met(O)-Thr(Msob)-Ser(Msob)-Glu(OMsob)-Lys(Msz)-Ser(Msob)-Gln-Thr(Msob)-Pro-Leu-Val-Thr(Msob)-Leu-O-DSB-resin] was deprotected and cleaved from the resin by the treatment with SiCl₄-thioanisole-anisole (100 eq. each)/TFA-dichloromethane (DCM) (9:1) at 25 °C for 3 h (the cleavage yield: 82 %). After purification of the crude γ -endorphin using reverse phase FPLC (fast protein liquid chromatography), the total yield based on the starting Boc-Leu-resin was 62 %. The product was identical with the authentic sample purchased from Peptide

Institute Inc. on reverse phase HPLC. Using a new His derivative, His(MsBom) (MsBom=4-methylsulfinylbenzyloxymethyl),⁷ we synthesized another model peptide, valosin (H-Val-Gln-Tyr-Pro-Val-Glu-His-Pro-Asp-Lys-Phe-Leu-Lys-Phe-Gly-Met-Thr-Pro-Ser-Lys-Gly-Val-Leu-Phe-Tyr-OH)¹⁴ in the same manner. In this synthesis, the total yield was 26 % and the cleavage yield was 70 %. The synthetic valosin was identical with the authentic sample purchased from Novabiochem on reverse phase HPLC.

REFERENCES AND NOTES

- Presented in part at the 29th Symposium on Peptide Chemistry, Oct. 24-26, 1991, Tokyo, Japan, Peptide Chemistry 1991, p. 187, and the 12th American Peptide Symposium, June 16-21, 1991, Cambridge Mass., U.S.A., Peptides: Chemistry and biology, p. 533.
- Y. Kiso, M. Shimokura, T. Kimura, T. Mimoto, M. Yoshida and T. Fujisaki, "Peptide Chemistry 1986", ed T. Miyazawa, Protein Res. Found., Osaka, Japan, 1987, p. 211.
- 3. J. M. Samanen and E. Brandies, J. Org. Chem., 1988, 53, 561.
- 4. S. Futaki, T. Taike, T. Akita and K. Kitagawa, J. Chem. Soc., Chem. Commun., 1990, 523.
- 5. Y. Kiso, T. Kimura, M. Yoshida, M. Shimokura, K. Akaji and T. Mimoto, J. Chem. Soc., Chem. Commun., 1989, 1511.
- 6. Y. Kiso, S. Tanaka, T. Kimura, H. Itoh and K. Akaji, Chem. Pharm. Bull., 1991, 39, 3097.
- T. Kimura, S. Tanaka, H. Itoh, Y. Fujiwara, K. Akaji and Y. Kiso, "Peptide Chemistry 1990", ed Y. Shimonishi, Protein Res. Found., Osaka, Japan, 1991, p. 5.
- 8. Y. Kiso, H. Itoh, S. Tanaka, T. Kimura and K. Akaji, Tetrahedron Lett., 1993, 34, 7599.
- Y. Kiso, M. Yoshida, T. Fujisaki, T. Mimoto, T. Kimura and M. Shimokura, "Peptide Chemistry 1986", ed T. Miyazawa, Protein Res. Found., Osaka, Japan, 1993, p. 205; Y. Kiso and T. Kimura, J. Synth. Org. Chem. Jpn., 1990, 48, 1046.
- 10. R. B. Merrifield, J. Am. Chem. Soc., 1964, 86, 304.
- 11. N. Nimura, H. Ogura and T. Kinoshita, J. Chromatogr., 1980, 202, 375.
- N. Ling, R. Burgus and R. Guillemin, Proc. Natl. Acad. Sci. USA, 1976, 73, 3942; N. Ling, Biochem. Biophys. Res. Commun., 1977, 74, 248.
- Y. Kiso, T. Kimura, Y. Fujiwara, H. Sakikawa and K. Akaji, Chem. Pharm. Bull., 1990, 38, 270;
 Y. Kiso, Y. Fujiwara, T. Kimura, A. Nishitani and K. Akaji, Int. J. Peptide & Protein Res., 1992, 40, 308.
- W. E. Schmidt, V. Mutt, M. Carlquist, H. Kratzin, J. M. Conlon and W. Creutzfeldt, FEBS Lett., 1985, 191, 264.

(Received in Japan 21 January 1994; accepted 23 February 1994)